

<b>SET</b>	<b>A/B/C</b>
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**INDIAN SCHOOL MUSCAT  
FINAL EXAMINATION 2023  
SUBJECT WITH SUBJECT CODE**

CLASS:

Max.Marks:

<b>MARKING SCHEME</b>			
<b>SET</b>	<b>Q.N. O</b>	<b>VALUE POINTS</b>	<b>MARKS SPLIT UP</b>
A	1.	d	1
	2.	B	1
	3.	c	1
	4.	c	1
	5.	b	1
	6.	a	1
	7.	b	1
	8.	d	1
	9.	c	1
	10.	b	1
	11.	a	1
	12.	d	1
	13.	b	1
	14.	b	1
	15.	A	1
	16.	A	1
	17.	C	1

	<b>18.</b> D	1
	<b>19.</b> EF is $\text{CH}_2\text{Cl}$ MF is $\text{C}_2\text{H}_4\text{Cl}_2$	1 1
	<b>20.</b> LR $\text{H}_2$ 27 g of $\text{H}_2\text{O}$ formed ,5 g of $\text{O}_2$ is in excess	1 1
	<b>21.</b> (a) Nitrogen has half-filled EC (b) Greater effective nuclear charge on $\text{Na}^+$ OR (a) Same order (b) Same order	$\frac{1}{2} + \frac{1}{2}$ 1 1 1
	<b>22.</b> MO Electronic Configuration, diamagnetic, $\text{BO} = 3$ OR (a) Resonance and delocalization of pi electrons (b) Any one point of difference between BE and BDE	1 $\frac{1}{2}$ $\frac{1}{2}$ 1 1
	<b>23.</b> $16\text{Zn} + 2\text{NO}_3^- + 10\text{H}^+ \rightarrow 8\text{Zn}^{2+} + \text{N}_2\text{O} + 5\text{H}_2\text{O}$	1+1
	<b>24.</b> (a) Correct bond line structure (b) Correct order and reason	1 1
	<b>25.</b> (a) 3-Ethylhex-3-ene (b) Yes aromatic , $(4n+2)\pi$ electrons	1 1
	<b>26.</b> $E = 4.97 \times 10^{-19} \text{ J}$ $\text{KE} = 0.972 \text{ eV}$ (or) $1.55 \times 10^{-19} \text{ J}$	1 1 1
	<b>27.</b> Oxdn half Reduction half $2\text{Cr}(\text{OH})_3 + \text{IO}_3^- + 4\text{OH}^- \rightarrow 2\text{CrO}_4^{2-} + 5\text{H}_2\text{O} + \text{I}^-$	1 1 1
	<b>28.</b> (a) HBr/Peroxide (b) $\text{H}_2\text{O}/\text{Hg}^{2+}, \text{H}^+$ (c) Zn OR (a) $\text{CH}\equiv\text{CNa}$ and $\text{NaC}\equiv\text{CNa}$ (b) $\text{CH}_3\text{C}(\text{CH}_3)(\text{OH})\text{CH}_3$ (c) $\text{CH}_2=\text{CH}_2$	1x3
	<b>29.</b> Correct equations for Wurtz reaction Friedel crafts alkylation Kolbes electrolysis	1x3
	<b>30.</b> (i) 18.4 M (ii) 0.0679 L (or) 67.9 mL	1 1/2 1 1/2

	<b>31.</b>	(a) $3p = 3+1 = 4$ (b) 5p, 4f, 6d (c) i) $m_l = 0$ , $l = 0$ ii) p subshell  OR $\Delta x = 9.59 \times 10^{-10} \text{ m}$	1 1 1 1 2
	<b>32.</b>	(a) Correct configuration d block elements (b) $Z = 39$ (c) 104, Unnilquadium  OR IE- A>C>B EGE- A>B>C	1 1 1+1 1 1
	<b>33.</b>	(a) $\text{NO}_2^-$ -Correct structure (b) FC = Zero (or) +2 (c) Any correct example (d) i) trigonal bipyramidal ii) Square planar  OR  (a) any two examples for expanded octet (b) one point of difference between BMO and ANMO (c) any two conditions (d) Sp <sup>3</sup> ,energy level diagram and orbital diagram	1 1 1 1 1 1 1 1+1
	<b>34.</b>	(a) I) Cyclohexene ii) propane-1,2-diol is formed iii) o and p chloroanisole (b) Ethyne>Ethene>Ethane -Difference in hybridization of C  OR  (a) Toluene, $\text{CH}_3$ is electron releasing (or) activating gp (b) But-1-yne will give white ppt with tollens reagent (c) Eclipsed & staggered conformations of ethane (d) 3-methylpent-2-ene	1 1 1 1+1 1 1 ½+½ 1
	<b>35.</b>	(a) Ether & Alcohol (b) 3-methylpent-3-en-2-ol (c) Correct structure (d) Definition of positive resonance effect Resonance structure of phenol	1 1 1 1 1
SET B			
	<b>1.</b>	b	1
	<b>3.</b>	d	1
	<b>7.</b>	d	

	<b>8</b>	c	
	<b>10</b>	a	
	<b>14</b>	a	
	<b>26</b>	(a) HBr/Markovnikovs addition (c) Fe/873K	
	<b>27</b>	(a) Decarboxylation (b) Wurtz reaction (c) Friedel crafts alkylation	
	<b>35</b>	a (iii) Benzenehexachloride (or) Hexachlorocyclohexane (or) Gammaxane	1
SET C	<b>2</b>	a	
	<b>3</b>	a	
	<b>5</b>	b	
	<b>7</b>	a	
	<b>10</b>	c	
	<b>19</b>	(a) 3-ethylpent-2-ene structure	1
	<b>27</b>	(b) O <sub>3</sub> /Zn/H <sub>2</sub> O (c) H <sub>2</sub> O/Hg <sup>2+</sup> /H <sup>+</sup>	1 1